

# INDOLE DERIVATIVES

## LXXXVIII.\* SYNTHESIS OF SOME O-( $\gamma$ -ALKOXY- $\beta$ -HYDROXYPROPYL)

### DERIVATIVES OF SEROTONIN

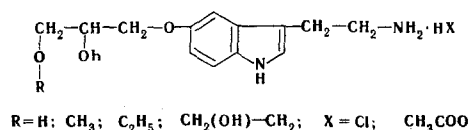
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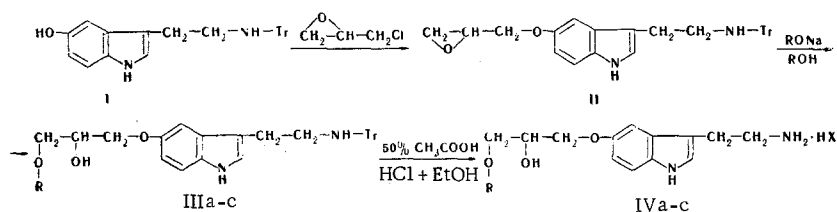
The reaction of N-tritylserotonin with epichlorohydrin has given O-( $\beta$ , $\gamma$ -epoxypropyl)-N-tritylserotonin, the reaction of which with sodium alkoxides in the corresponding alcohols has given O-( $\gamma$ -alkoxy- $\beta$ -hydroxypropyl) derivatives of serotonin. It has been found that they possess a definite radioprotective action.

Serotonin possesses a considerable radioprotective activity [2]. Serotonin derivatives with an ester linkage (O-acyl and O-carbamoyl compounds) do not differ substantially in relation to the nature and duration of the radioprotective effect from serotonin itself. The length of the O-acyl residue has little influence on the radioprotective properties of these compounds [3]. For serotonin derivatives with an ether linkage (alkoxytryptamines), in contrast to the O-acyl derivatives, lengthening the hydrocarbon chain in position 5 of the indole ring of an alkoxytryptamine leads to an increase in the toxicity of the materials and to a loss of protective properties [4].

In view of the above facts, it appeared of interest to obtain some O-alkyl derivatives of serotonin containing functional groups in the side chain with the general formula



The presence in the serotonin molecule of two reactive groups ( $\text{NH}_2$  and  $\text{OH}$ ) complicated the problem of introducing the appropriate groupings in position 5. Consequently, we used triphenylmethyl (trityl) protection of the amino group which, after the synthesis of the appropriate compounds, was removed under conditions not affecting the groupings introduced into position 5 of the indole ring. As the starting material we selected 5-benzyloxytryptamine, which is an intermediate in the synthesis of serotonin. After the protection of the amino group of the 5-benzyloxytryptamine, the benzyl residue was eliminated from the hydroxy group. The following route for the synthesis was selected:



\* For Communication LXXXVII, see [1].

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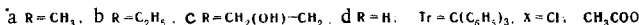


TABLE 1. Antiradiation Efficiency of the Materials on Parenteral Administration

$$\begin{array}{c} \text{CH}_2-\text{CH}-\text{CH}_2-\text{O}-\text{C}_6\text{H}_4-\text{CH}_2-\text{CH}_2-\text{NH}_2\cdot\text{HCl} \\ | \quad | \\ \text{OR} \quad \text{OH} \end{array}$$

Compound	R	Dose, mmole/kg	No. of mice	Survival rate, %	MLT, days
IVd	H	0.4	20	60.0±10.9	9.6±1.7
		0.15	10	60.0±15.5	11.0±3.7
IVa	CH <sub>3</sub>	0.4	10	30.0±14.5	8.5±1.8
IVb*	C <sub>2</sub> H <sub>5</sub>	0.4	10	50.0±15.8	9.2±2.0
IVc	HOC <sub>2</sub> H <sub>4</sub>	0.4	10	0	8.2±1.3
Tryptamine hydrochloride		0.4	20	55.0±11.1	14.3±1.9
		0.15	30	36.6± 8.8	13.7±1.4
5-Methoxytryptamine (mexamine) hydrochloride		0.4	30	53.0± 9.1	14.2±2.3
		0.15	29	89.4± 5.7	18.6±4.1
Control group		—	34	8.9± 4.9	10.6±0.6

\*Tested in the form of the acetate.

TABLE 2. Constants and Elementary Analyses of the Compounds of the General Formula

$$\begin{array}{c} \text{CH}_2-\text{CH}-\text{CH}_2-\text{O}-\text{C}_6\text{H}_4-\text{CH}_2-\text{CH}_2-\text{NH}_2\cdot\text{HX} \\ | \quad | \\ \text{O} \quad \text{OH} \\ | \\ \text{R} \end{array}$$

Compound	R	mp, °C	Empirical formula	Found, %				Calc., %				Yield, %
				C	H	N	Cl	C	H	N	Cl	
IVa	CH <sub>3</sub>	192—194 (dec.)	C <sub>14</sub> H <sub>21</sub> N <sub>2</sub> O <sub>3</sub> Cl <sub>1</sub>	56.0	7.0	9.2	11.7	55.9	7.0	9.3	11.8	83
IVb	C <sub>2</sub> H <sub>5</sub>	137	C <sub>17</sub> H <sub>26</sub> N <sub>2</sub> O <sub>3</sub>	60.2	7.8	8.3	—	60.3	7.7	8.3	—	85
IVc	CH <sub>2</sub> OHCH <sub>2</sub>	149—150	C <sub>15</sub> H <sub>23</sub> N <sub>2</sub> O <sub>4</sub> Cl <sub>1</sub>	54.2	7.2	8.4	10.6	54.4	7.1	8.5	10.7	80
IVd	H	156—157	C <sub>13</sub> H <sub>19</sub> N <sub>2</sub> O <sub>3</sub> Cl <sub>1</sub>	54.5	6.7	9.6	12.5	54.4	6.7	9.8	12.4	67

$\alpha$ -CH<sub>2</sub>), 4.18 (q,  $\alpha$ -CH<sub>2</sub>), J<sub>gem</sub>=11.1; J<sub>vic</sub>=6.2; J<sub>vic</sub>=3.0; 3.28 (m,  $\beta$ -CH); 2.5–2.9 (m,  $\gamma$ -CH<sub>2</sub>), J<sub>gem</sub>=5.5, J<sub>vic</sub>=2.2, J<sub>vic</sub>=3.8. IR spectrum: 918 cm<sup>-1</sup> (epoxide ring).

O-( $\beta$ -Hydroxy- $\gamma$ -methoxypropyl)-N-tritylserotonin (IIIa). A mixture of 2.84 g (6 mmoles) of (II) and a solution of 1.5 mmole of CH<sub>3</sub>ONa in 50 ml of absolute methanol was boiled in a current of nitrogen for 10 h. After the end of the reaction, the solution was evaporated in vacuum. On cooling, the residue deposited crystals of O-( $\beta$ -hydroxy- $\gamma$ -methoxypropyl)-N-tritylserotonin. The reaction product was purified on a column of alumina and was recrystallized from methanol. This gave 2.68 (86%) of O-( $\beta$ -hydroxy- $\gamma$ -methoxypropyl)-N-tritylserotonin with mp 141–143°C. Found: C 78.1; H 6.8; N 5.6%. C<sub>33</sub>H<sub>34</sub>N<sub>2</sub>O<sub>3</sub>. Calculated: C 78.2; H 6.8; N 5.5%. PMR spectrum: 3.25 (s, CH<sub>3</sub>O); 5.08 (d, OH), J=5.8 Hz.

O-( $\gamma$ -Ethoxy- $\beta$ -hydroxypropyl)-N-tritylserotonin (IIIb). A mixture of 2.37 g (5 mmoles) of (II) and 1.25 mmole of C<sub>2</sub>H<sub>5</sub>ONa in 20 ml of absolute ethanol was boiled in a current of nitrogen for 5 h. After the end of the reaction the solution was evaporated and cooled. The crystals of (IIIb) that deposited were filtered off. The reaction product was purified on a column of alumina and was recrystallized from ethanol to give 2.2 g (84%) of (IIIb) with mp 69–70°C. Found: C 78.4; H 7.0; N 5.4%. C<sub>34</sub>H<sub>36</sub>N<sub>2</sub>O<sub>3</sub>. Calculated: C 78.5; H 7.0; N 5.4%. PMR spectrum: 3.67–4.10 (m,  $\alpha$ ,  $\beta$ -CH<sub>2</sub>-CH); 1.11 (t, CH<sub>3</sub>); 5.03 (d, OH), J=4.6 Hz.

O-( $\beta$ -Hydroxy- $\gamma$ -hydroxyethoxypropyl)-N-tritylserotonin (IIIc). A mixture of 3.32 g (7 mmoles) of (II) and 1.75 mmole of sodium glycolate in 30 ml of ethylene glycol was stirred in a current of nitrogen at a bath temperature of 120–125°C for 5 h. After the end of the reaction, the solution was cooled, and distilled water was added. The white flocculant precipitate of (IIIc) that deposited was filtered off, dried, and freed from impurities on a column of alumina. A solution of (IIIc) in chloroform was evaporated to dryness in vacuum and the residue was recrystallized from methanol to give 3.08 g (82%) of product with mp 84–86°C. Found: C 76.2; H 6.7; N 5.3%. C<sub>34</sub>H<sub>36</sub>N<sub>2</sub>O<sub>4</sub>. Calculated: C 76.1; H 6.8; N 5.2%. PMR spectrum:

3.88 (m,  $\alpha$ ,  $\beta$ -CH<sub>2</sub>-CH); 4.56 (t, OH on a primary carbon atom), J=5.2 Hz; 5.07 (d, OH on a secondary carbon atom), J=4.7 Hz.

O-( $\beta$ , $\gamma$ -Dihydroxypropyl)-N-tritylserotonin (III<sub>d</sub>). A mixture of 1.25 g (3 mmoles) of N-tritylserotonin, 0.28 ml (3.35 mmoles) of glycerol monochlorohydrin, 0.42 g (3 mmoles) of potassium carbonate, 0.45 g (3 mmoles) of sodium iodide, and 6.5 ml of methyl ethyl ketone was boiled in a current of nitrogen for 25 h. After the end of the reaction, the inorganic salts were filtered off and the mother solution was evaporated to dryness in vacuum. The reaction product was chromatographed on a column of alumina with chloroform-methanol (10:1) as the eluent. The solution containing the (III<sub>d</sub>) was evaporated to dryness in vacuum, and the residue was crystallized from ethanol. This gave 0.62 g (43%) of (III<sub>d</sub>), mp 147-148°C. Found: C 78.1; H 6.6; N 5.7%. C<sub>32</sub>H<sub>32</sub>N<sub>2</sub>O<sub>3</sub>. Calculated: C 78.0; H 6.5; N 5.6%. PMR spectrum: 4.05 (m,  $\alpha$ ,  $\beta$ -CH<sub>2</sub>-CH); 3.71 (d,  $\beta$ -CH); 4.61 (t, OH on a primary carbon atom), J=6.5 Hz; 4.87 (d, OH on a secondary carbon atom), J=6.5 Hz.

Removal of the Trityl Protection. To convert the compound (III<sub>a-d</sub>) into the form of salts (acetate or hydrochloride), 1.25 g of one of these substances was suspended in 32 ml of 50% acetic acid. The suspension was heated in a current of nitrogen with stirring up to a bath temperature of 140-150°C and was then stirred at this temperature for 5 min, after which the solution was cooled and the crystals of triphenylmethanol were filtered off. The mother solution was evaporated to dryness in vacuum and was dissolved in absolute methanol. In the preparation of the acetate of (IV<sub>b</sub>), dry ether was added to a solution of this salt and the mixture was cooled. The crystals of the acetate of (IV<sub>b</sub>) that separated out were filtered off and dried. In the preparations of compounds (IV<sub>a</sub>, c, d) in the form of the hydrochlorides, an excess of a titrated ethanolic solution of hydrogen chloride (1.1 mole of HCl per mole of substance) was added to an ethanolic solution of the acetate, and the solution was evaporated to dryness in vacuum. The dry residue was dissolved in 2-3 ml of absolute methanol, dry ether was added until the solution became turbid, and the mixture was cooled. The crystals that deposited were filtered off and dried. The constants and elementary analyses of the compounds obtained are given in Table 2.

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